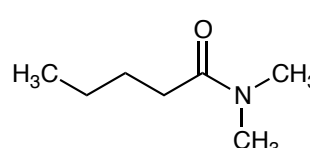
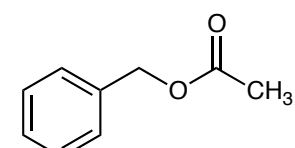
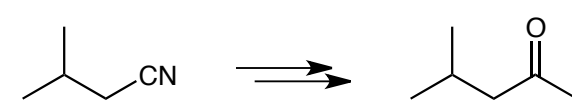
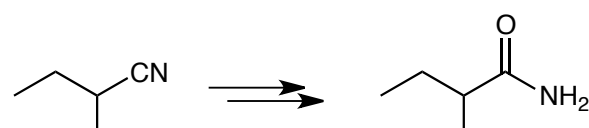
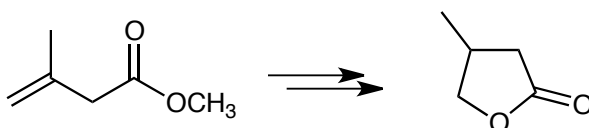
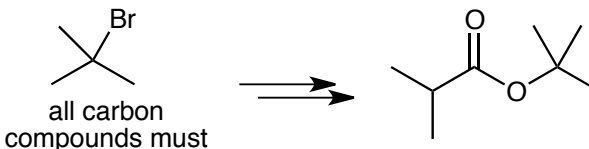
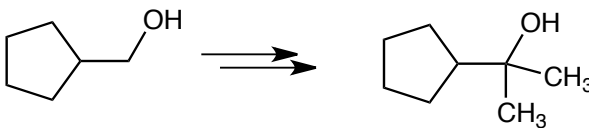
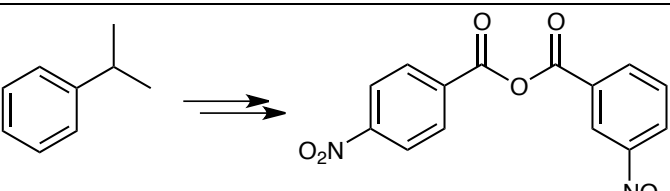
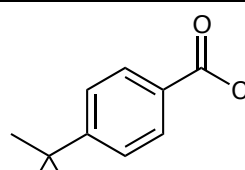
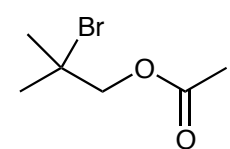
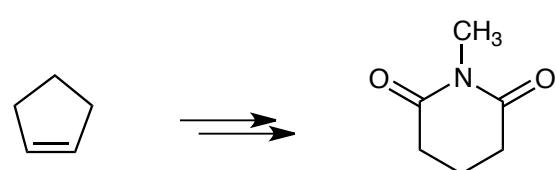
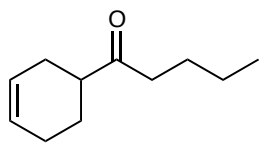
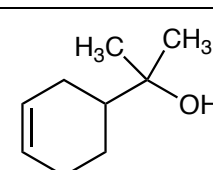
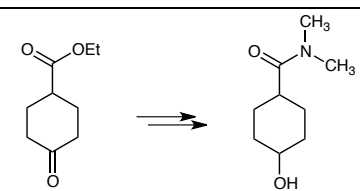
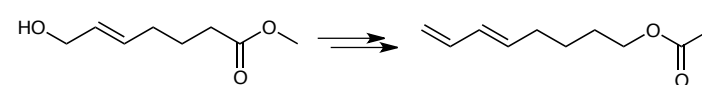
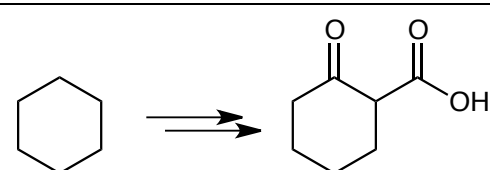


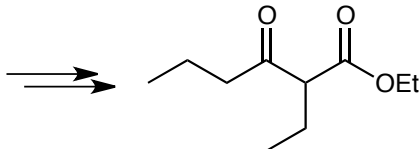
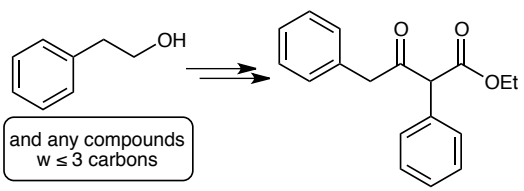
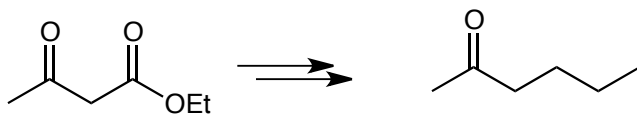
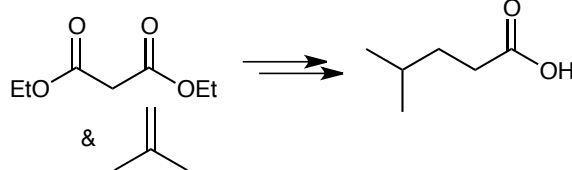
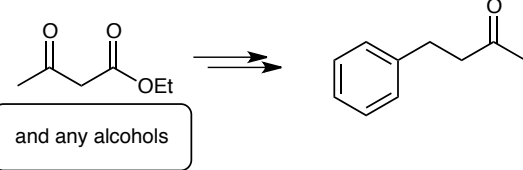
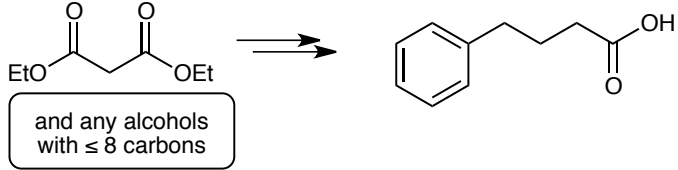
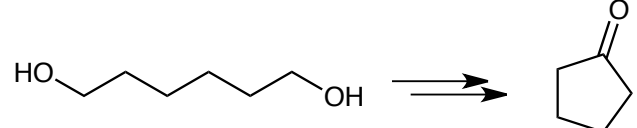
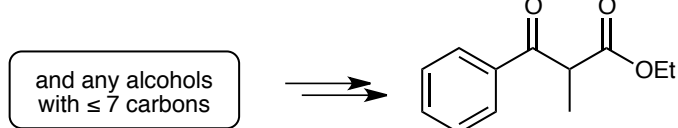
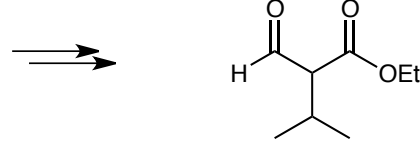
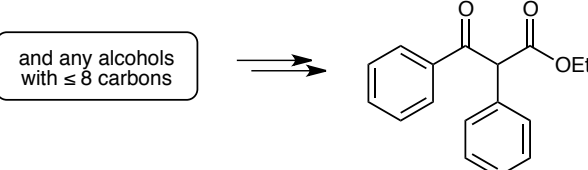
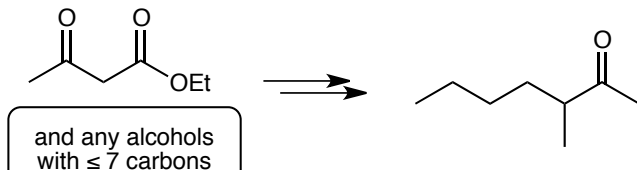
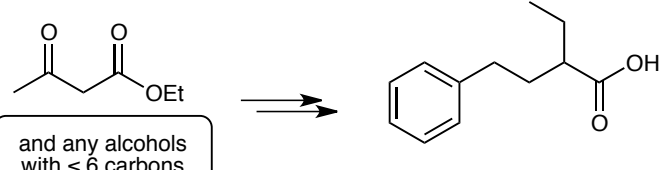
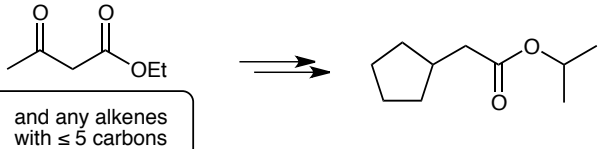
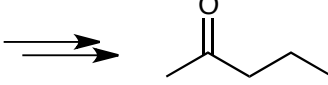
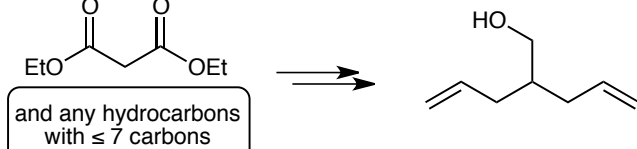
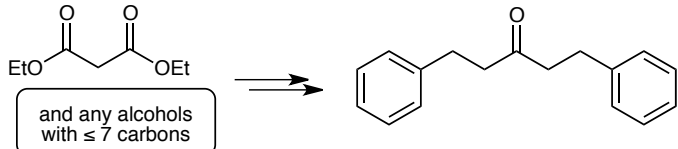
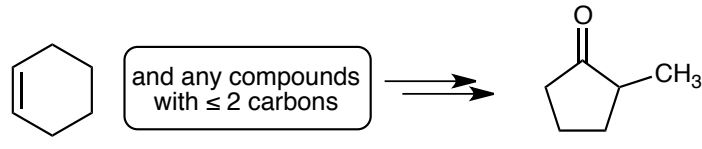
## CHAPTER 21: CARBOXYLIC ACID DERIVATIVES

The key step in all of the syntheses below is nucleophilic acyl substitution to form a carboxylic acid derivative. Redox reactions of carbonyls and alcohols are also featured. Grignard additions to nitriles and esters is a common method for chain extension.

<div style="border: 1px solid black; padding: 5px; display: inline-block;">any alcohols &amp; any amines</div> $\Rightarrow$ 	<div style="border: 1px solid black; padding: 5px; display: inline-block;">any alcohols</div> $\Rightarrow$ 
	
	 <p>all carbon compounds must begin with this</p>
	
<div style="border: 1px solid black; padding: 5px; display: inline-block;">hydrocarbons w/ <math>\leq 6</math> carbons</div> $\Rightarrow$ 	<div style="border: 1px solid black; padding: 5px; display: inline-block;">any alkenes</div> $\Rightarrow$ 
	<div style="border: 1px solid black; padding: 5px; display: inline-block;">any compounds with <math>\leq 4</math> carbons</div> $\Rightarrow$ 
<div style="border: 1px solid black; padding: 5px; display: inline-block;">any compounds with <math>\leq 4</math> carbons</div> $\Rightarrow$ 	
	

## CHAPTER 22 & 23: CONDENSATION REACTIONS OF ESTER ENOLATES

Condensation of ester enolates (Claisen) is the key reaction in most of these syntheses. Acetyl ketones are prepared by the acetoacetic ester synthesis. Substituted acetic acids are prepared by the malonic ester synthesis.

<p>any alcohols with <math>\leq 5</math> carbons</p> 	 <p>and any compounds with <math>\leq 3</math> carbons</p>
	
 <p>and any alcohols</p>	 <p>and any alcohols with <math>\leq 8</math> carbons</p>
	 <p>and any alcohols with <math>\leq 7</math> carbons</p>
<p>and any alcohols with <math>\leq 7</math> carbons</p> 	 <p>and any alcohols with <math>\leq 8</math> carbons</p>
 <p>and any alcohols with <math>\leq 7</math> carbons</p>	 <p>and any alcohols with <math>\leq 6</math> carbons</p>
 <p>and any alkenes with <math>\leq 5</math> carbons</p>	<p><math>\text{H}_2\text{C}=\text{CH}_2</math></p> <p>all carbon compounds must begin with this</p> 
 <p>and any hydrocarbons with <math>\leq 7</math> carbons</p>	 <p>and any alcohols with <math>\leq 7</math> carbons</p>
 <p>and any compounds with <math>\leq 2</math> carbons</p>	

## CHAPTER 22 & 23: ALPHA SUBSTITUTION REACTIONS; ALDOL CONDENSATION

Condensation of aldehydes (Aldol) gives  $\beta$ -hydroxycarbonyls and  $\alpha,\beta$ -unsaturated carbonyls. Conjugate (1,4) addition of nucleophiles to  $\alpha,\beta$ -unsaturated carbonyls is a key step in several syntheses. Substitution reactions at the  $\alpha$ -carbon (e.g., Hell-Volhard-Zelinski and Haloform) are also featured.

